

Oxidation - the process of recoil electrons an atom, molecule or ion.

- Atoms are converted into positively charged ion:
$$\text{Zn}^0 - 2e \rightarrow \text{Zn}^{2+}$$
- negatively charged ion becomes neutral atom:
$$2\text{Cl}^- - 2e \rightarrow \text{Cl}_2^0$$
$$\text{S}^{2-} - 2e \rightarrow \text{S}^0$$
- The value of the positively charged ion (an atom) is increased accordingly the number of electron donating:

